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1-PHENYLHEPTANE-1,5-DIONE FROM PHELLINUS TREMULUS

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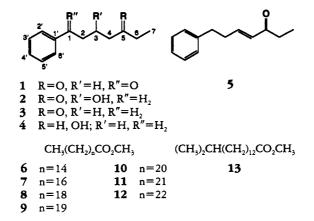
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ABSTRACT.—1-Phenylheptane-1,5-dione [1], which has not been previously reported as a natural product, has been isolated from *Phellinus tremulus*. This compound is the first naturally occurring phenylheptane oxidized in the 1 position and the first dione of this class of natural products.

Phellinus tremulus Bondarzew et Borissov (Polyporaceae) is a fungus responsible for heart rot in aspen (Populus tremuloides) (1). The fungus, which is rich in melanin pigments (2), gradually renders the infected wood unsuitable for pulping due to low levels of essential carbohydrates (3). The phenylheptanes 2-5 have been isolated from the decayed heartwood of aspens infected with the organism (4). We report herein the isolation of the dione 1 directly from the fungus. This is the first report of this compound as a natural product, the first report of the isolation of any phenylheptane natural product directly from the fungus rather than the infected wood, the first report of a phenylheptanedione, and the first report of a phenylheptane oxidized at the 1 position isolated from this fungus.

Compound 1 was isolated from the hexane-soluble fractions of the MeOH extract of *Pb. tremulus* by chromatography on Si gel using EtOAc in hexane. Elution first afforded the known esters 6-13 and the phenylheptanone 3, after which the new phenylheptanedione 1 eluted as a pale yellow oil which crystallized on standing. Long-chain fatty acids similar to those isolated from *Ph. tremulus* occur in several species of fungus and plants, including *Fomes igniarius* (5), *Agaricus bisporus* (6), and pecans (*Carya illinoensis*) (7).

The mass spectrum of the new natural product exhibited a molecular ion at m/z 204 and a base peak at m/z 105, consistent with the molecular formula as $C_{13}H_{16}O_2$ and the presence of a C_6H_5CO moiety. The ¹H (Table 1) and ¹³C nmr (Experimental) revealed the presence of two ketonic carbonyl groups (one of which is conjugated), a monosubstituted aromatic ring, an isolated ethyl group, and a trimethylene group. The ir spectrum showed carbonyl absorptions at 1684 and 1717 cm⁻¹. These data clearly estab-



| Proton | Chemical Shift | Multiplicity | Coupling Constants |
|------------------|----------------|--------------|---|
| H-7 | 1.02 | t | $J_{67} = 7.35$ |
| н-6 | 2.40 | q | $J_{47} = 7.35$ |
| H-4 | 2.51 | t | $J_{3} = 7.14$ |
| H-3 | 1.98 | toft | $J_{1,4} = 7.14, J_{2,3} = 6.96$ |
| H-2 | 2.98 | t | $I_{2,3} = 6.96$ |
| H-2', H-6' | 7.93 | d | $J_{6,7} = 7.35$ $J_{6,7} = 7.35$ $J_{3,4} = 7.14$ $J_{3,4} = 7.14, J_{2,3} = 6.96$ $J_{2,3} = 6.96$ $J_{2',3'} = J_{5',6'} = 7.40$ |
| H-3', H-4', H-5' | 7.6-7.36 | m | <u> </u> |

TABLE 1. ¹H-nmr Spectral Data (200 MHz) for 1.

lished the constitution of the natural product as 1. Comparison of the observed ¹H-nmr spectral data with those published (8) for 1 confirmed the assignments.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.— Nmr spectra were obtained using a Varian Gemini 200 spectrometer at 200 MHz (¹H) and 50 MHz (¹³C) using CDCl₃ solutions with TMS as standard. Ms were obtained using a Hewlett-Packard 5970 MSD gc-ms system containing a 30 m capillary column with a 5% phenylsilicone/95% methylsilicone liquid phase with oven temperatures being programmed at 100° for 3 min, then ramped at 10°/min for 18 min; ionization was carried out at 70 eV. Ir spectra were recorded by diffuse reflectance from KBr using a BioRad FTS 60-A ir spectrophotometer.

FUNGAL MATERIAL.—*Pb. tremulus* was collected in the Snake River State Forest in central Minnesota in April and May 1992. A voucher specimen is preserved in the Department of Chemistry, South Dakota State University, Brookings.

ISOLATION OF COMPOUND 1.—Dried fungus (11.3 kg) was extracted three times with MeOH (30 liters) for a week, and the combined extracts were evaporated under reduced pressure to yield 163 g of oil. The crude extract (67 g) was partitioned between $H_2O-CH_2Cl_2(1:1)$ to yield 27 g of a CH_2Cl_2 -soluble fraction; further partitioning of this fraction between 10% $H_2O/MeOH$ and hexane yielded 11.0 g of a hexane-soluble fraction. This fraction (5.0 g) was subjected to cc on Si gel (125 g adsorbent; EtOAc in hexane eluent). The material eluted from the column with 20% EtOAc in hexane was subjected to preparative tlc (Si gel, 20% EtOAc/hexane); the fractions of R_f 0.6–0.7 were subjected to preparative hplc (Si gel, 10% EtOAc/hexane) and afforded **1** as a pale yellow oil (18 mg) which crystallized on standing. Recrystallization from C₆H₆/hexane afforded **1** as a white crystalline solid: mp 62–64°; ν max cm⁻¹ (KBr) 3068, 2970, 2942, 1717, 1684, 1456, 1387, 1279, 1218, 1122, 755, 706 cm⁻¹; λ max nm (log ϵ) (EtOH) 238 (4.04); ¹³C nmr δ 211.8 (C-3), 200.2 (C-1), 137.5 (C-1'), 133.5 (C-4'), 129.0, 128.5 (C-2', C-3'), 41.7, 37.8 (C-2, C-4), 36.0 (C-6), 18.5 (C-3), 8.0 (C-7) ppm; ¹H nmr see Table 1; eims *m/z* (%) [**M**]⁺ 204 (9), 175 (8), 148 (4), 147 (21), 120 (29), 105 (100), 77 (46).

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